5-Acyl-4-amino-2-phenyl-1,3-oxazin-6-ones

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Intramolecular cyclization of acyl(ethoxycarbonyl)ketene N-benzoylaminals in boiling Ph₂O gives 5-acyl-4-amino-2-phenyl-1,3-oxazin-6-ones.

Key words: intramolecular cyclization, acyl(ethoxycarbonyl)ketene *N*-benzoylaminals, 5-acyl-4-amino-2-phenyl-1,3-oxazin-6-ones, ¹⁷O NMR.

Azoles and azines containing vicinal amino and acyl groups are convenient building blocks for synthesizing condensed nitrogen-containing systems. We have previously found convenient methods for the synthesis of new reagents of this type belonging to triazole, pyridine, and pyrimidine to triazole, pyridine, and pyrimidine acyle from acyle, diacyle, and acyle (ethoxycarbonyl) ketene aminals or N,S-acetals containing one or two unsubstituted NH₂ groups. 7–9

In the present work we studied one of the possible approaches to the synthesis of substituted 1,3-oxazines from ketene aminals, which are readily obtained from alkyl β-ketocarboxylates and benzoylcyanamide in the presence of catalytic amounts of Ni(acac)₂.8 We have found that refluxing of acyl(ethoxycarbonyl)ketene N-benzoylaminals (1a,b) in Ph₂O results in intramolecular cyclization with elimination of EtOH to give crystalline products, which, according to the spectral data (IR, ¹H, ¹³C, and ¹⁵N NMR), are 4-amino-5-acyl-2-phenyl-1,3-oxazin-6-ones (2a,b).

The alternative structure of 1*H*-azet-2-one derivatives (3a,b) was unambiguously ruled out based on ¹⁷O NMR data recorded in CD₃CN. For example, the spectrum of the product obtained from 1a displays signals at 235, 308, and 487 ppm. The first should obviously be attributed ¹⁰ to the O(1) atom of the oxazine

ring in compound **2a**. However, this signal cannot be ascribed to any of the O atoms of the carbonyl groups in compound **3a**.

The yields of oxazinones **2a,b** are as high as 60—67 %. According to IR spectroscopic data (in CH₂Cl₂) and ¹H NMR spectra (in CDCl₃) for these compounds, one of the H atoms of the amino group is involved in an N—H···O=CR intramolecular hydrogen bond.

Previously, ¹¹ acylation of diaminomethylenemalonic ester and ethyl diaminomethylenecyanoacetate with PhCOCl gave heterocyclization products, which were regarded to be the corresponding derivatives of 1*H*-azet-2-one (*cf.* 3a,b). The structure of these compounds seems to require additional consideration.

4-Amino-2-aryl-1,3-oxazin-6-ones substituted at the exocyclic N atom have been obtained¹² by thermolysis of 2-(N-aroyl)diaminomethylene derivatives of Meldrum's acid. However, the possibility of an azetone-type alternative structure was not discussed.

In addition, the interest in the chemistry of 1,3-oxazinones is, to a considerable extent, determined by their ability to undergo various transformations during nucleophilic attack of the ring (for reviews see Refs. 13 and 14). Therefore, compounds 2a,b can be used both in reactions involving annelation of the pyridine or pyrimidine ring and in reactions with ring transformations.

Experimental

¹H NMR spectra were recorded on a Bruker WM-250 spectrometer. ¹³C, ¹⁵N, and ¹⁷O NMR spectra were obtained on a Bruker AM-300 instrument. IR spectra were recorded on an UR-20 spectrophotometer. Mass spectra were measured on a Varian MAT-311A mass spectrometer (EI, 70 eV). Acyl(ethoxycarbonyl)ketene aminals **1a,b** were synthesized according to a known procedure.⁸

Intramolecular cyclization of keteneaminals 1a,b (general procedure). A solution of keteneaminal 1a,b (0.01 mol) in Ph₂O (25 mL) was refluxed for 1.5 h under Ar and cooled to

~20 °C. Hexane (60 mL) was added, and the resulting precipitate was filtered off and purified by column chromatography (CHCl₃/CCl₄ 1:1, SiO₂) to give oxazinones **2a,b**, which were then recrystallized from benzene.

4-Amino-5-acetyl-2-phenyl-1,3-oxazin-6-one (2a). Yield 67 %, m.p. 165-166 °C (benzene). Found (%): C, 62.67; H, 4.40; N, 12.19. $C_{12}H_{10}N_2O_3$. Calculated (%): C, 62.60; H, 4.38; N, 12.17. MS, m/z: 230 [M]⁺⁻. IR (CH₂Cl₂), v/cm^{-1} : 3465, 3250 br (NH); 1765, 1740 (C=O), 1636, 1615, 1596, 1580. ¹H NMR (CDCl₃), δ: 2.67 (s, 3 H, Me); 6.21 (s, 1 H, NH); 7.50–8.30 (m, 5 H, Ph); 10.45 (s, 1 H, NH). ¹³C NMR (CDCl₃), δ: 31.99 (q, Me, J=129 Hz); 90.94 (s, C(5)); 128.95, 129.0, 129.3, 134.4 (Ph); 159.14 (s, C(4)); 164.91 (s, C(6)); 165.29 (t, C(2), J=3 Hz); 199.27 (q, CO, J=7 Hz); ^{15}N NMR (DMSO-d₆), δ: $^{-1}63.4$ (d, C=N, J=7 Hz); $^{-2}68.70$ (t, NH₂, J=91 Hz). ^{17}O NMR (CD₃CN), δ: 235.00 (O—C=O), 308.00 (O—C=O), 487.00 (MeCO).

4-Amino-5-benzoyl-2-phenyl-1,3-oxazin-6-one (2b). Yield 60 %, m.p. 185–187 °C (benzene). Found (%): C, 69.93; H, 4.09; N, 9.71. $C_{17}H_{12}N_2O_3$. Calculated (%): C, 69.85; H, 4.14; N, 9.59. MS, m/z: 292 [M]⁺⁺. IR (CH₂Cl₂), v/cm^{-1} : 3470, 3275 br (NH), 1760 sh, 1745 (CO), 1628, 1598, 1580. ¹H NMR, CDCl₃, δ: 9.90 (s, 1 H, NH), 7.30–8.40 (m, 10 H, 2 Ph), 6.23 (s, 1 H, NH). ¹³C NMR (CDCl₃), δ: 89.58 (s, C(5)), 127.85, 128.92, 129.74, 131.31, 134.39, 140.63 (2 Ph), 158.60 (s, C(4)), 165.26 (s, C(6)), 165.74 (s, C(2)), 195.88 (CO).

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References

- 1. M. F. Gordeev, A. V. Komkov, V. S. Bogdanov, and V. A. Dorokhov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1990, 1392 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1990, **39**, 1256 (Engl. Transl.)].
- 2. V. A. Dorokhov, A. V. Komkov, and B. I. Ugrak, Izv.

- Akad. Nauk, Ser. Khim., 1993, 1429 [Russ. Chem. Bull., 1993, 42, 1364 (Engl. Transl.)].
- V. L. Gein, S. G. Pitirimova, O. V. Vinokurova, Yu. S. Andreichikov, A. V. Komkov, V. S. Bogdanov, and V. A. Dorokhov, *Izv. Akad. Nauk*, Ser. Khim., 1994, 1475 [Russ. Chem. Bull., 1994, 43, 1398 (Engl. Transl.)].
- 4. V. A. Dorokhov, M. F. Gordeev, A. V. Komkov, and V. S. Bogdanov, *Izv. Akad. Nauk, Ser. Khim.*, 1990, 145 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1990, 39, 130 (Engl. Transl.)].
- V. A. Dorokhov, M. F. Gordeev, A. V. Komkov, and V. S. Bogdanov, Izv. Akad. Nauk, Ser. Khim., 1991, 159 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1991, 40, 142 (Engl. Transl.)].
- A. V. Komkov, B. I. Ugrak, V. S. Bogdanov, and V. A. Dorokhov, *Izv. Akad. Nauk*, *Ser. Khim.*, 1994, 1469 [*Russ. Chem. Bull.*, 1994, 43, 1392 (Engl. Transl.)].
- V. A. Dorokhov, M. F. Gordeev, E. M. Shashkova, A. V. Komkov, and V. S. Bogdanov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1991, 2600 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1991, 40, 2274 (Engl. Transl.)].
- 8. V. A. Dorokhov, M. F. Gordeev, Z. K. Dem'yanets, M. N. Bochkareva, and V. S. Bogdanov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1989, 1806 [Bull. Acad. Sci. USSR, Div. Sci., 1989, 38, 1654 (Engl. Transl.)].
- 9. V. A. Dorokhov, M. F. Gordeev, A. V. Komkov, and V. S. Bogdanov, *Izv. Akad. Nauk SSSR, Ser. Khim.*, 1990, 401 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1990, 39, 340 (Engl. Transl.)].
- D. W. Boykin, D. W. Sullins, N. Pourahmady, and E. J. Eisenbraun, *Heterocycles*, 1989, 29, 301.
- N. D. Bodnarchuk and A. A. Yatsishin, Zh. Org. Khim., 1977, 13, 954 [J. Org. Chem., 1977 (Engl. Transl.)].
- 12. Xian Huang, Bang-Chi Chen, Guo-Yong Wu, and Hong-Bo Chen, Synth. Comm., 1991, 21, 1213.
- Y. Yamamoto and Y. Morita, J. Synth. Org. Chem., Japan, 1992, 50, 887.
- 14. V. E. Zakhs, I. P. Yakovlev, and B. A. Ivin, Khim. Geterotsikl. Soedin., 1987, 1443 [Chem. Heterocycl. Comp., 1987 (Engl. Transl.)].

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